Iterative Benzyne–Furan Cycloaddition **Reactions: Studies toward the Total** Synthesis of ent-Sch 47554 and ent-Sch 47555

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ABSTRACT



7-Fluoro-5,8-dimethoxy-1-naphthol, prepared from the lithiation and benzyne formation from 1,4-difluoro-2,5-dimethoxybenzene and Diels-Alder cycloaddition with furan, was sequentially C-glycosidated under Suzuki conditions and O-glycosidated using di-O-acetyl-L-rhamnal to provide the corresponding β -naphthyl C,O-disaccharide. Further lithiation, benzyne formation, and cycloaddition with furan gave an oxabridged 1,4-dihydroanthracenyl C,O-disaccharide, a model compound relevant to the total synthesis of Sch 47555.

Sch 47554 (1) and Sch 47555 (2) (Figure 1) are two angucycline antibiotics1 that were isolated at Schering-Plough from the fermentation broths of a strain of Streptomyces sp. (SCC-2136), originally isolated from a Canadian soil sample.² Both compounds showed antifungal activities against a variety of yeasts and dermatophytes including Candida albicans, Candida tropicalis, Candida stellatoidea, Trichophyton mentagrophytes, Trichophyton rubrum, Trichophyton tonsurans, and Microsporum canis with the greater activity shown by Sch 47554 (1).² As part of our ongoing interest in diverse biologically active benz[a] anthracene antibiotics,³ we wished to explore the use of 1,4-difluoro-2,5-dimethoxybenzene (7), an equivalent of the double benzyne 1,4dimethoxycyclohexa-1,2,3-trien-5-yne,⁴ in the synthesis of model anthraquinones structurally related to the ring ABC unit of Sch 47554 (1).

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Figure 1. Sch 47554 (1) and Sch 47555 (2).

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The synthesis of *C*-aryl glycosides from the Lewis acid catalyzed rearrangement of the corresponding *O*-glycosides is well documented as a powerful method for their construction from simple precursors in a regio- and stereocontrolled manner when the glycoside is ortho to a phenol or naphthol unit.^{5–8} Lewis acids including boron trifluoride etherate,^{9–14} tin tetrachloride,¹⁵ scandium triflate,¹⁶ trimethylsilyl triflate,^{17,18} and Cp₂MCl₂–AgClO₄ (M = Zr, Hf)^{19–24} have been used to catalyze the reaction. In this rearrangement, the choice of Lewis acid is critical to the yield and the stereoselectivity of the transformation.

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To construct the A–C ring unit of Sch 47554 (1), we sought to utilize O-glycosidation of naphthol 6, O to C transglycosidation, and subsequent further glycosidation via a Ferrier-type rearrangement²⁵ using glycal **4**. We have previously reported the synthesis of naphthol 6 from 1,4-difluoro-2,5-dimethoxybenzene (7) via a benzyne-furan Diels-Alder reaction (Scheme 1).⁴ However, the O to C trans-glycosidation does pose several potential problems. It is known that the rearrangement of a naphthol with a halide substituent at C-3 is inefficient and only provides C-glycosides in very poor yields (<5%).^{11,26,27} These results likely arise through perturbation by the halide substituent reducing electron density at the position ortho to the naphthol. Nonetheless, Suzuki has described the use of a catalytic quantity of scandium triflate¹⁶ or excess hafnocene dichloride and silver perchlorate²¹ to mediate the rearrangement of halo-substituted O-glycosides to provide C-glycosides in good to excellent yields. As convenient model studies, we sought to prepare the naphthyl disaccharide 8 (Figure 8) and related compounds



Figure 2. Target model disaccharide 8.

corresponding to the enantiomers of the A-C ring units of the natural products 1 and 2.

The glycosyl donor *ent-5* was synthesized from di-*O*-acetyl-L-rhamnal $(4)^{28}$ following reaction with water at 80 °C¹¹ to provide the *trans*-alkene 9,²⁹ which was directly hydrogenated at ambient pressure to provide the lactol 10³⁰ (Scheme 2). Subsequent acetylation gave the diacetate *ent*-5. Scandium triflate or hafnocene dichloride and silver perchlorate promoted condensation of the aryl fluoride 6 and acetate *ent*-5 gave the desired equatorial *C*-aryl glycoside 11 (44% or 59% respectively) as a single diastereoisomer (Scheme 2). *O*-Methylation proceeded smoothly to provide naphthalene 12, which was saponified using methanolic potassium carbonate to give alcohol 13. Ferrier-type rearrangement of di-*O*-acetyl-D-rhamnal³¹ (*ent*-4) in the presence of alcohol 13 promoted by indium trichloride³² gave acetate

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14 (78%). The full structure and relative stereochemistry of *C*-glycoside 14 was confirmed by X-ray crystallographic structure determination. Subsequently enone 16 was prepared in two steps (69%) by hydrolysis of acetate 14 using potassium carbonate in methanol followed by immediate oxidation of the unstable alcohol 15 using manganese dioxide.

Hydrogenation of alkene 14 and subsequent saponification gave the ring A–C unit of *ent*-Sch 47555 18 (90%). In contrast saponification of 14 followed by hydrogenation resulted in extensive decomposition. To examine benzyne formation, alcohol 18 was protected as the benzyl ether 8 (83%) and allowed to react with *n*-butyllithium and furan. This gave the cycloadduct 19 (85%) as a 1:1 mixture of diastereoisomers (Scheme 3). Cycloadduct 19 was allowed to react with acetic acid in tetrahydrofuran at reflux to give the *C*-aryl glycoside 20 (66%). Clearly, the disaccharide unit proved to be more prone to cleavage than the oxa-bridged 1,4-dihydroanthracenyl unit under acidic conditions.



In summary, we have successfully prepared enantiomeric model ring A–C disaccharide units **16** and **8** for Sch 47554 (**1**) and Sch 47555 (**2**) from difluoride **7** using *O* to *C trans*-glycosidation and a Ferrier-type *O*-glycosidation. In addition, the fluoride **8** was converted into the corresponding benzyne, which was trapped with furan in a Diels–Alder reaction. Further investigations into the total synthesis of Sch 47554 (**1**) and Sch 47555 (**2**) will be reported in due course.

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Supporting Information Available: Experimental procedures and structural data for all new compounds; X-ray crystallographic structure for **14** (CIF); copies of ¹H NMR, ¹³C NMR, and NOESY NMR spectra for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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